



# Knight Piésold

Co-Disposal Commingling

Paste Rock

### DEFINITION

Mine waste streams are typically separated according to their particle size (due to their origin in the mining process)

Conventionally disposed of separately

Co-disposal combines waste streams in any of a variety of ways

### CO-DISPOSAL WITH MINING PRODUCTS

- Tailings: disposed as a slurry
  - high porosity
  - water-filled voids.
- Waste Rock:
  - high porosity (>30%)
  - air-filled voids "hopefully"

**Co-disposal -** tailings filling at least some of the voids in the coarse waste.

### TYPICAL "RULE OF THUMB" POROSITIES

	Conv. Slurried Tailings	Compacted/ Filtered Tailings	Waste Rock
Particle SG	2.65	2.65	2.65
Dry Density (pcf)	65 – 90	90 – 110	90 – 110
Void Ratio	0.8 – 1.5	0.5 – 0.8	0.5 – 0.8
Porosity, pct	50 – 60	35 – 45	35 – 50

Porosity: ratio of volume of voids to total volume (\*100%) Void ratio: ratio of volume of voids to volume of solids

The intent is to "hide" tailings within the voids of the waste rock

### EXAMPLE EFFECT OF C:F RATIO



Orroad of worde filled with toilinge



### **DEGREES OF MIXING**

929

Wickland et al.

Table 1. Methods of co-disposal.

Description

Homogeneous mixtures: waste rock and tailings are blended to form a homogeneous mass (placement method unknown)

Pumped co-disposal: coarse and fine materials are pumped to impoundments for disposal (segregation occurs)

Layered co-mingling: alternating layers of waste rock and tailings

Waste rock is added to a tailings impoundment

Tailings are added to a waste rock dump

Waste rock and tailings are disposed in the same depression

Separate disposal: waste rock in dumps and tailings in impoundments

## <u>PERFECTLY</u> BLENDED, HOMOGENEOUS MIXTURES OF TAILINGS AND WASTE ROCK: PASTE ROCK "Paste rock"

any mining companies have found that the "mine waste" issue is growing and won't go away. In the previous "Tailings Tips" article (CMJ August 2007), Golder Associates' Don Welch observed that simply finding an acceptable place to put tailings and waste rock is becoming a challenge. Then there are the long-term liability issues associated with conventional tailings facilities, and the question of who will be around to maintain these structures a century or so from now.

Finding a better way to store tailings so that the metals, acids and other hazards they may contain stay isolated from the rest of the environment, goes a long way to reducing costs and liabilities.

One possible answer to these challenges is to dispose



Co-author Ward Wilson stands on a paste rock deposition, supported by the rock matrix under the paste.

### BLENDED, RELATIVELY HOMOGENEOUS MIXTURES OF TAILINGS AND WASTE ROCK

- Tailings and mine waste are mixed relatively homogeneously
- Tailings essentially fill "all" the voids between waste rock particles
- Waste rock has predominantly rock-torock contact

### Blended Tailings/Waste Rock

### Benefits/Goals:

- Reduced footprint
- Shear strength like waste rock
- Permeability like tailings
- Low oxygen diffusion rates
- Greatly(?) reduced ARD potential
- Improved permitting timeline
- Better public acceptance
- Improved closure opportunities...

### Governing equations for ARD

The common chemical equations associated with ARD are shown below:

$$2 \operatorname{FeS}_{2} + 7 \operatorname{O}_{2} + 2 \operatorname{H}_{2} \operatorname{O} \Leftrightarrow 2 \operatorname{Fe}_{2} + 4 \operatorname{SO}_{4}^{2-} + 4 \operatorname{H}^{+}$$
(1)

$$2 \operatorname{Fe}^{2+} + 1/2 \operatorname{O}_{2} + 2 \operatorname{H}^{+} \Leftrightarrow 2 \operatorname{Fe}^{3+} + \operatorname{H}_{2} \operatorname{O}$$
(2)

$$2 \operatorname{Fe}^{3+} + 6 \operatorname{H}_{2} O \Leftrightarrow 2 \operatorname{Fe}(OH)_{3} + 6 \operatorname{H}^{+}$$

$$FeS_{2} + 14 \operatorname{Fe}^{3+} + 8 \operatorname{H}_{2} O \Leftrightarrow$$

$$15 \operatorname{Fe}^{2+} + 2 \operatorname{SO}_{4}^{2-} + 16 \operatorname{H}^{+}$$
(3)
(4)

There is a similar reaction for carbonate rocks:

 $CaCO_{3} + 2 H^{+} + SO_{4}^{2-} \Leftrightarrow$   $CaSO_{4} + H_{2}O + CO_{2}$ (5)

Keep the Fe from going ferric

### Above about 85% Saturation: Oxygen Diffusion Essentially Eliminated

Fig. 2. Comparison between diffusion coefficient values measured on different materials (soils, tailings, and geosynthetic clay liners; data taken from Aubertin et al. 1999, 2000*b*; and Aachib et al. 2002<sup>4</sup>) at various  $S_r$ , with predicted values obtained with the model of Collin (1987) and the proposed eq. [16].



From: Mbonimpa et al., 2003, Figure 2.

Keeping the air out will keep the Fe from going ferric



Proxy waste rock



### LA Abrasion Machine

# 11 Secret Herbs and Spices



### PROCTOR TEST RESULTS



### Proctor in terms of solids content



### At least in these tests: a lower solids content is better



### Zero Air Voids





6-inch dia. Triaxial shear testing

3:1 Blend



4:1 Blend



# **Co-disposed Tailings**

Challenges:

- Maintaining +/- proper mixing ratios
- Defining proper mixing ratios
- Developing design properties (hydraulic, shear strength, etc.)
- Mixing method
- Placement/spreading method

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# Material Mixing: Where to from Here?





Figure 2 Concrete transit mixer truck and skip



Figure 3. Skip during column loading

# No, no, bigger, BIGGER!

### SECRECTS TO MIXING???



### Flow-through Ball Mill



### Agglomerator

Continuous concrete mixerPUG millAgglomerator

They all want very dilute slurry and small sized rock particles

# Material Mixing: The Concept















![](_page_33_Picture_0.jpeg)

![](_page_33_Picture_1.jpeg)

![](_page_34_Figure_0.jpeg)

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